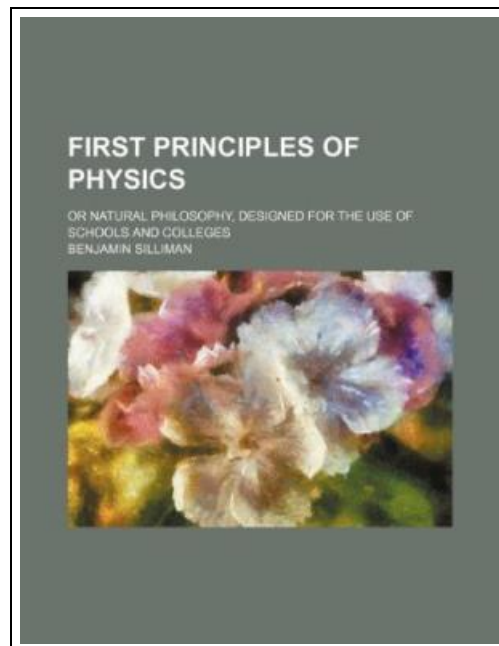


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1859 Excerpt: .heat of aeriform bodies, like that of solids and liquids, increases by condensation and diminishes by rarefaction. The continued diminution in temperature as we ascend in the atmosphere, is due chiefly to this cause. The average diminution in temperature in ascending from the sea level is 1F. for every 300 feet Supposing the average temperature of the air at the level of the sea, near the equator, to be 80, and toward the poles 0, the figures in the second and third column of the following table will express approximately the temperature at different elevations. Decrease of temperature in the atmosphere from elevation. 603. Specific heat of the same body in a liquid or solid state. A body in the liquid state has a greater specific heat than when it is in the solid form, as might naturally be concluded from the fact, that the addition of heat is necessary to convert the solid into a liquid. Thus ice has a specific heat of 0-505, water being 1000, sulphur (solid,) 0-2026--fluid, 0-2340; phosphorus, between 45 and--6, 0-1887, at 212, 0-2045, Ac, The high specific heat of water moderates very greatly the rapidity of natural transitions from heat to cold and from cold to heat, owing to the large quantity of heat emitted or absorbed by the ocean in accommodating itself to variations in external temperature. 601. How does mechanical compression affect specific heat? 602. How do condensation and the reverse affect the specific heat of...

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