



## A Pharmaceutical Guide to the First and Second Examinations

By John Barker Smith

Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1870 Excerpt: . to a moderate heat until the nitric oxide be driven off. Magnesium, light; Magnesium, Zinc, Lime.--The light carbonate of the first and the carbonates of the others are heated to a dull red heat in a loosely-covered crucible until there is no effervescence when mixed with acids. Lime is mostly obtained by calcination of limestone on a large scale. 2. Precipitated oxides. Antimony.--A solution of the chloride is partly decomposed by addition of water, and finally by sodium carbonate, the precipitated oxide is then washed with boiling water and dried at a heat not exceeding 212. Iron, magnetic; Iron, peroxide, moist; Iron, hydrated peroxide.--The magnetic is formed on boiling a solution of iron sulphate and persulphate with soda solution; it must be dried at a heat less than 120: the humid peroxide is precipitated with soda solution, but without heat and from a solution of...



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