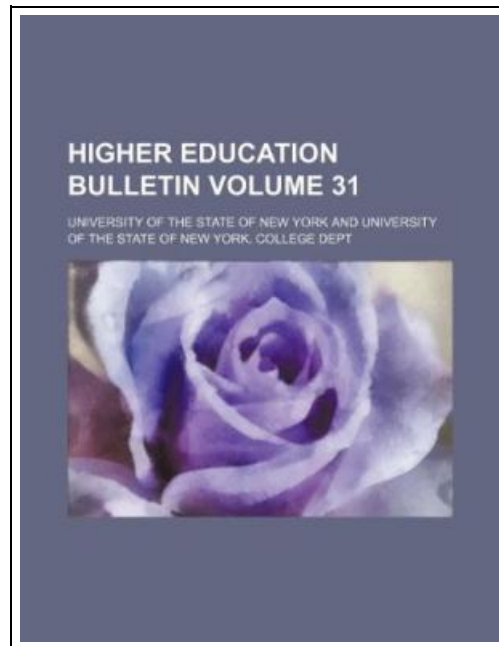


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1907 Excerpt: . 49TH Veterinary Examination Place CHEMISTRY Wednesday, June 26, 1907--9.15 a. m. to 12.15 p. m., only Answer any 10 of the questions on this paper but no more. Check (//) the number of each one of the questions you have answered. 1 Define the following terms: element, alloy, compound, mixture, solution, valence. 2 Complete the following equation and explain fully what takes place: $\text{CaCO}_3 + 2\text{HCl} =$ 3 Describe the properties and state the medicinal uses of the substance formed when hydrogen and oxygen are united in equal volumes. 4 How is chlorin prepared from sodium chlorid? Give the physical and the chemical properties of chlorin. 5 What temperature Fahrenheit is equivalent to a temperature of 28 centigrade? What temperature centigrade is equivalent to a temperature of 120o Fahrenheit? 6 Mention the compounds symbolized as follows: HgCl_2 ; KNO_2 ; KNO_3 ; SnCl_2 ; CO ; N_2O ; NaHSO_4 ; $\text{Fe}_2(\text{SO}_4)_3$. 7 State the difference between metals and nonmetals. Name three of each. 8 Complete the following equations: $2\text{NaOH} + \text{H}_2\text{SO}_4 = 2\text{NaCl} + \text{H}_2\text{SO}_4 = \text{CaCO}_3 + \text{heat} = \text{Fe} + 2\text{HCl} =$ 9 Give the chemical reaction of (1) blood, (2) gastric juice, (3) urine, (4) sweat. 10 Give the formula and the chemical name of each of the following substances: aqua fortis, epsom salts, washing soda, saltpeter, blue stone. 11 Mention the different varieties of amorphous carbon and give the properties of each. 12 Give the graphic formula of (1) marsh gas, (2) water, (3) sulfuric acid, (4) stannic chlorid. 13 How much oxygen can be obtained...



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