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By William Macgill

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Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1872 Excerpt: . Thus, if we plunge 2 lbs. of iron at 200 into 3 lbs. of water at 50--required the temperature. The spec fie heat of water is nine times that of iron, and $9 \times 3 = 27$. Then, $--27 \times 50 = 1350$ $2 \times 200 = 400$ $29 \ 29 \ 1750(60i$ very nearly. In questions under (6.), first work as before, then multiply or divide the answer by the same number, according as it is a substance of less or greater specific heat that is wanted. Thus we have 6 lbs. of lead at 250; what weight of water at 70 poured over it will reduce the whole to 90 I 250 present temperature. 90 90 proposed Water added 70 difference. 20 6 20 960 48 But the specific heat of water is 30 times as great; hence it will take 3D times less water, and--48 lbs.--30...

Reviews

Extensive manual! Its this type of great read through. This can be for all who statte there was not a worth reading. It is extremely difficult to leave it before concluding, once you begin to read the book.

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It becomes an remarkable publication that I have possibly go through. Better then never, though i am quite late in start reading this one. I am just delighted to inform you that this is basically the best ebook we have study inside my individual existence and can be he greatest book for actually.

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